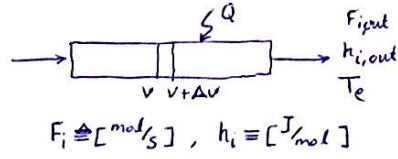


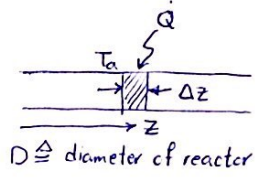
① Ideal Non-Isothermal PFRs

- Consider the general EBE $\frac{dE_{sys}}{dt} = \dot{Q} - \dot{W}_s + \sum_i F_{i,in} h_{i,in} - \sum_i F_{i,out} h_{i,out}$
 In most instances \dot{W}_s term can be neglected. $h_{i,o}$ and $h_{i,e}$ are specific enthalpy [$\frac{J}{mol}$] at inlet and exit conditions resp.
 Also for SS op: $\frac{dE_{sys}}{dt} = 0$
 hence, the EBE for the entire reactor is given by: $\dot{Q} + \sum_i F_{i,o} h_{i,o} - \sum_i F_{i,e} h_{i,e} = 0$ (I)
 however, F_i and h_i vary since comp and T vary along the length of PFR as the rxn proceeds.

- general EBE on a ΔV element would be: $\Delta \dot{Q} + (\sum_i F_i h_i)_V - (\sum_i F_i h_i)_{V+\Delta V} = 0$ (II)
 eq. II in terms of specific enthalpy on a mass basis $F_i h_{i,in} = m_i h'_{i,in}$
 (II) $\Rightarrow \Delta \dot{Q} + (\sum_i m_i h'_i)_V - (\sum_i m_i h'_i)_{V+\Delta V} = 0$ (III)
 let h' be the specific enthalpy of the reacting mixture (mass basis). and it is defined as: $h' = \frac{\sum_i m_i h_i}{\sum_i m_i}$ (IV)
 (IV) into (III) $\Rightarrow \Delta \dot{Q} + [(\sum_i m_i) h']_V - [(\sum_i m_i) h']_{V+\Delta V} = 0$ (V)
 but $\sum_i m_i = m \triangleq$ total mass flow rate and m remains the same due to conservation of mass.
 hence, (V) $\Rightarrow \Delta \dot{Q} + m(h'|_V - h'|_{V+\Delta V}) = 0$ (VI)



- The $\Delta \dot{Q}$ Term
 Let z be distance along the length of PFR
 $\Delta \dot{Q} = U_c (\Delta A) (T_a - T)$
 $\Delta A = \pi D \Delta z$, $\Delta V = \frac{\pi D^2}{4} \Delta z \Rightarrow \Delta z = \frac{4 \Delta V}{\pi D^2}$
 $\Rightarrow \Delta \dot{Q} = \frac{4 U_c}{D} \Delta V (T_a - T)$



- back to (VI) $\Rightarrow m(h'|_{V+\Delta V} - h'|_V) = \frac{4 U_c}{D} \Delta V (T_a - T)$
 taking the limit as $\Delta V \rightarrow 0 \Rightarrow m \frac{dh'}{dV} = \frac{4 U_c}{D} (T_a - T)$ (VII)

- Specific Enthalpy
 Due to sensible heat: $m \frac{dh'_{thermal}}{dV} = m C_p' \frac{dT}{dV}$ C_p' is on a mass basis
 Due to energy released or consumed in chemical rxn: $m \frac{dh'_{chemical}}{dV} = (\Delta H_{R,A}) (-r_A)$
 hence, eq. VII becomes: $m C_p' \frac{dT}{dV} + (\Delta H_{R,A}) (-r_A) = \frac{4 U_c}{D} (T_a - T)$
 $\Rightarrow \frac{dT}{dV} = \frac{4 U_c}{m C_p' D} (T_a - T) - \frac{(\Delta H_{R,A}) (-r_A)}{m C_p'}$ (VIII)
 or if the heat capacities are expressed in a molar basis: $\frac{dT}{dV} = \frac{4 U_c}{D} (T_a - T) - \frac{(\Delta H_{R,A}) (-r_A)}{\sum_i F_i C_{p,i}}$ (IX)

- The comp. balance eq. for a PFR is given by: $\frac{dX_A}{dV} = \frac{1}{F_{A0}} (-r_A)$ (X)

- To obtain T and X_A as a function of PFR volume, we must solve the above system of ODEs simultaneously.
 - Note that $(-r_A)$ must be expressed in terms of X_A and T
 $C_{p,i}$, $\Delta H_{R,A}$ ~ ~ ~ as a function of T.
 F_i as functions of X_A

- Ex. 01 "Non-Isothermal PFR"
 gas phase rxn, ideal gases, $\text{CH}_4 + \text{Cl}_2 \rightarrow \text{CH}_3\text{Cl} + \text{HCl}$
 PFR: $D = 7.5 \text{ cm}$, $V_R = 2 \text{ m}^3$, $F_{A0} = 24 \text{ mol/s}$
 $T_a = T_c$ (inlet Feed T), $\Delta H_{R,298} = -23 \text{ kcal/mol}$
 T and X_B profiles for $T_0 = 530 \text{ K}$, U_0 between 0-30 $\text{cal/m}^2\cdot\text{s}$
 $U_0 = 30 \text{ cal/m}^2\cdot\text{s}$, T_0 between 530-540 K

$-r_A = -r_B = k C_A C_B = 7.5 \times 10^{11} \frac{-17940}{T} C_A C_B \text{ [mol/m}^3\cdot\text{s]}$
 $F_{B0} = 6 \text{ mol/s}$, $P_0 = 0.2 \text{ MPa}$ (constant)
 $C_{P,A} = 17.10 \text{ cal/mol}\cdot\text{K}$, $C_{P,B} = 8.75 \text{ cal/mol}\cdot\text{K}$
 $C_{P,C} = 0.01 \text{ cal/mol}\cdot\text{K}$, $C_{P,D} = 7.07 \text{ cal/mol}\cdot\text{K}$

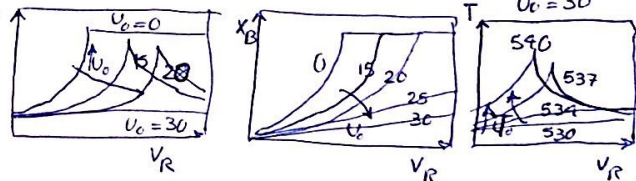
- Ex. 01 solution "MBE"

Comp. B is the limiting reactant $\Rightarrow -r_B = -\frac{dF_B}{dV} = F_{B0} \frac{dX_B}{dV} = k C_A C_B$
 stoichiometric table:
 $C_A = \frac{F_A}{F_t} C_t = \frac{F_{A0} - F_{B0} X_B}{F_{A0} + F_{B0}} \times \frac{P_0}{RT}$
 $C_B = \frac{F_{B0}(1-X_B)}{F_{A0} + F_{B0}} \times \frac{P_0}{RT}$
 thus, $\frac{dX_B}{dV} = 7.5 \times 10^{11} \exp\left(\frac{-17940}{T}\right) \frac{(F_{A0} - X_B F_{B0})(1-X_B)}{(F_{A0} + F_{B0})^2} \left(\frac{P_0}{RT}\right)^2$
 $F_t = F_{A0} + F_{B0}$

- Ex. 01 solution "EBE"

$\frac{dT}{dV} = \frac{4U_0/D (T_a - T) - (\Delta H_{R,B})(-r_B)}{\sum F_i C_{p,i}}$
 $\Delta H_{R,B} = \Delta H_{R,298}^\circ + \int_{298}^T (\Delta C_p) dT \Rightarrow \Delta H_{R,B} = \Delta H_{R,298}^\circ + \Delta C_p (T - 298)$
 $\sum F_i C_{p,i} = (F_{A0} - X_B F_{B0}) C_{p,A} + F_{B0}(1-X_B) C_{p,B} + F_{B0} X_B (C_{p,C} + C_{p,D})$
 finally: $\frac{dT}{dV} = \frac{4U_0/D (T_0 - T) - [\Delta H_{R,298}^\circ + \Delta C_p (T - 298)] \times 7.5 \times 10^{11} \exp\left(\frac{-17940}{T}\right) \frac{F_{B0}(1-X_B)(F_{A0} - X_B F_{B0})}{(F_{A0} + F_{B0})^2} \left(\frac{P_0}{RT}\right)^2}{(F_{A0} - X_B F_{B0}) C_{p,A} + F_{B0}(1-X_B) C_{p,B} + F_{B0} X_B (C_{p,C} + C_{p,D})}$

numerical solution using Runge-Kutta 4th order with T following initial conditions: @ $V_R = 0$; $T = T_0$, $X_B = 0$



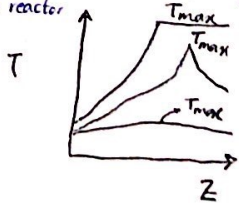
② Parametric Sensitivity of PFRs

- In the previous example for a Non-Isothermal PFR

we saw extreme responses in the T and Concentration profiles resulting from small changes in operating conditions. These types of extreme responses of a PFR occur for highly exothermic rxns when the reactor is operating under certain conditions. Such behavior is termed parametric sensitivity. Obviously, the most important problem associated with parametric sensitivity is to determine conditions under which this sensitivity is to be expected and what criteria may be established to ensure its absence. Such dramatic rise in T may \rightarrow destroy catalyst which are T-sensitive. \rightarrow causes fatigue in the reactor construction material. \rightarrow lead to explosion.

- Accurrence of "Hot spots" in PFRs

"EBE" in a PFR:
 for a tubular PFR: $dV = \frac{\pi D^2}{4} dz$ ②, where z is the distance along the reactor.
 ② in ① $\Rightarrow \frac{dT}{dz} = \frac{\pi D U_0}{4 m C_p} (T_a - T) - \frac{(\Delta H_{R,A})(-r_A)}{m C_p} \left(\frac{\pi D^2}{4}\right)$ ③ \rightarrow describes the variation in rxn T along the length of the reactor
 The "Hot-spot" T is defined as the max reactor T.
 T_{max} occurs when: $\frac{dT}{dz} = 0$
 hence, at the hot-spot T, we have: $\frac{4U_0}{D} (T_a - T_{max}) = (\Delta H_{R,A})(-r_A)$ ④



heat removed by heat transfer to surroundings at T_a heat generated by the rxn

- To establish some control over sensitivity

we can specify some allowable T_{max} and manipulate heat transfer coefficient U_0 and/or reactor wall T (T_a) such that at T_{max} : rate of generation of heat \leq rate of removal of heat (5)

suppose at the hot-spot T, T_{max} , the reactor concentration is $C_{A,max}$.

we can rewrite eq. 5 as: $\frac{4U_0}{D}(T_a - T_{max}) = (\Delta H_{R,A})[-r_A(T_{max}, C_{A,max})]$ (6)

The problem is $C_{A,max}$ is not known a priori. rxn rate @ T_{max} and corresponding to $C_{A,max}$

To obtain $C_{A,max}$ we must solve the ODEs of energy and material balance.

However, we can make a conservative estimate: \rightarrow reactor inlet concentration

• eq. 7 is true for powerlaw kinetics ($-r_A = kC_A^n$) $-r_A(T_{max}, C_{A,max}) < -r_A(T_{max}, C_{A0})$ (7)

(6), (7) $\Rightarrow \frac{4U_0}{D}(T_a - T_{max}) \leq (\Delta H_{R,A})[-r_A(T_{max}, C_{A0})]$ (8)

hence, to avoid parametric sensitivity, we can specify T_{max} and manipulate U_0 and/or T_a such that the above inequality holds.

plot \rightarrow Note that the two curves are obtained by specifying a T_{max} and evaluating $-r_A$.

$-r_A(T_{max}, C_{A0})$ is known since we know C_{A0} and specified T_{max} , but $-r_A(T_{max}, C_{A,max})$ is not known a priori.

The solid straight line's $\begin{cases} \text{intercept} = T_a \\ \text{slope} = 4U_0/D \\ \text{and corresponds to the heat removal} \end{cases}$

$T_{max,2}$ is the design specified max T based on $-r_A(T_{max}, C_{A0})$ curve.

the actual max T that will be obtained, however, is $T_{max,1}$.

